

WHITTEN, DAVIS & PECK

General Chemistry, 6/e
 General Chemistry with Qualitative Analysis, 6/e

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Chapter 17 Errata, first print run only

Chapter 17 Chemical Equilibrium experienced the following problem due to a production error during the first print run on July 31, 1999: Several exponents and subscripts appear on the baseline of an equation and could be misconstrued. A second print completed in November 1999 and all subsequent print runs of this edition will show the correct format.

Page	Correct format	Correct expression
711 middle	c is subscript in K_c	K_c
716	c, d, a, b are exponents	$\frac{[C]^c[D]^d}{[A]^a[B]^b}$
721 bottom	x and y are exponents	$\text{Rate}_f = k_f[A]^x[B]^y$
733 above Example 17-13	c, d, a, b are exponents	$K_P = \frac{(P_C)^c(P_D)^d}{(P_A)^a(P_B)^b}$
734 next to margin note	n is exponent next to delta	$K_P = K_c(RT)^{\Delta n}$ or $K_c = K_P(RT)^{-\Delta n}$
734 bottom	n is exponent next to delta	$K_P = K_c(RT)^{\Delta n}$
736 margin note	n is exponent next to delta	$K_P = K_c(RT)^{\Delta n}$
739	c, d, a, b are exponents	$K = \frac{(a_C)^c(a_D)^d}{(a_A)^a(a_B)^b}$

740 above table	c, d, a, b are exponents	$[C]^c[D]^d \dots = [A]^a[B]^b \dots$
740 table	c, d, a, b are exponents	$[C]^c[D]^d \dots = [A]^a[B]^b \dots$
742 bottom	Where K_{T_1} and K_{T_2} appear: T_1 is subscript of K and T_2 is subscript of K	K_{T_2} K_{T_1}
743 entire page	Where K_{T_1} and K_{T_2} appear: T_1 is subscript of K and T_2 is subscript of K	K_{T_2} K_{T_1}
743 margin note	n is exponent next to delta	$K_c = K_p(RT)^{-\Delta n}$
744 top right	c, d, a, b are exponents	$[C]^c[D]^d$ <hr/> $[A]^a[B]^b$